## **Exercice IV**

1) Which of the following compounds is an oxidant:

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B_2O_3 \\ Na_2B_4O_7 \\ NaBO_3(H_2O)_4 \\ Na_3BO_3
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- 2) What are the two 'tricks' which are used to produce Aluminum by electrolysis?
- 3) The +1 oxidation state is particularly important for which element: Al, B, Tl or Ga?
- 4) In which reaction is the MX<sub>3</sub> compound not behaving as a Lewis acid:

5) The boron atoms in borazine are susceptible to:

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attack by a nucleophile attack by an electrophile
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